



RESEARCH ARTICLE

Development and validation of an UV-Vis spectrophotometric method for the quantification of oclacitinib in capsule formulation

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Abstract

Objective. The aim of this study was the development and validation of an UV-Vis spectrophotometric method for the quantification of oclacitinib in commercial capsule formulation since pharmacopeias have not yet provided an official monograph for this drug.

Methods. The parameters linearity, limit of detection, limit of quantitation, specificity, precision, accuracy, and robustness were determined according to Brazilian and international guidelines.

Results. Linearity was determined for the analytical range of 5-15 µg/mL, and a limit of detection of 1.18 µg/mL and limit of quantification of 3.58 µg/mL were obtained. The method was selective and the precision was demonstrated through repeatability and intermediate precision, with relative standard deviations of 1.96% and 1.78%, respectively. In its turn, accuracy presented recovery percentages of 98.32-100.91%. All robustness and sample stability (48 h at 25 °C) results revealed no statistical variation among the groups. **Conclusions.** The presented method is suitable for the quantification of oclacitinib in commercial capsule formulation.

Keywords: Analytical validation. Oclacitinib. UV-Vis spectrophotometric.

How to cite

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INTRODUCTION

Oclacitinib (OCL) or trans-N-methyl-4-(methyl-7H-pyrrolo[2,3-d]pyrimidin-4-ylamino)-cyclohexanemethanesulfonamide (Figure 1) is a Janus Kinase (JAK) inhibitor (JAK1 preferentially) that blocks the signaling of proinflammatory cytokines. It is used for allergic responses and other hypersensitive diseases, such as pruritus and atopic dermatitis (AD) in dogs and cats¹. This substance is currently not approved for use in humans, but it is available for veterinary use and is provided as coated tablets or capsule formulations from compounding pharmacies¹.

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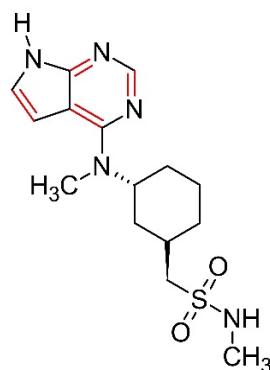


Figure 1. Chemical structure of OCL. Conjugated double bonds are highlighted in red.

AD has a prevalence of 3-15% in the dog population and represent 3-58% of dogs with diagnosed skin disease²⁻⁴. The pathogenesis of this disease is complex and involves genetic and environmental factors. The infiltration of the skin by inflammatory cells and the production of cytokines due to allergens from food and microbial or insect sources is one of the mechanisms of AD⁵. It is a life-long disease that can be controlled, but rarely cured. Therefore, rational clinical management of AD is necessary⁶.

OCL has a good safety profile in the treatment of AD, in addition to effectiveness and fast action⁷. The initial dosage is 0.4-0.6 mg/kg taken orally twice a day, but after two weeks it can be reduced to once daily⁸⁻¹⁰. The adverse effects are uncommon, but include anorexia, vomiting and diarrhea⁸⁻¹².

The quality control of medicines involves a series of procedures that are carried out to ensure that the products are not released for commercialization or distribution, until their quality is considered satisfactory. Among the tests performed, quantification of the drug in the pharmaceutical dosage form stands out^{13,14}.

Although OCL is already used in clinical practice, there are no data on this drug in official monographs, and there are no methods for the quantification of OCL in pharmaceutical dosage forms. Thus, there is a need to develop an accurate, precise and robust method for a suitable quality control in manipulated formulas. The present work reports the development and validation of an UV-Vis spectrophotometric method for the quantification of OCL in commercial capsule formulation manipulated.

MATERIAL AND METHODS

Chemicals

OCL capsules (5 mg), OCL maleate (Cayman Chemicals, Michigan, USA; standard reference), and a matrix (mixture of excipients) were kindly donated by a local compounding pharmacy (Curitiba, Brazil). Ultrapure water was obtained from a Milli-Q® Gradient A10 purification system (Millipore, Milford, USA). High-performance liquid chromatography grade methanol was purchased from Honeywell Riedel-de-Haën (Seelze, Germany) and Tedia (Fairfield, USA).

Equipment and instrumental conditions

An UV-Vis spectrophotometer Agilent 8453 (Santa Clara, USA) and a quartz cuvette (with an optical path length of 1 cm) were used to perform all absorbance analyses at 287 nm. The spectral range was from 200 nm to 900 nm.

Standard solutions and calibration curve

A stock solution was prepared in triplicate at 1 mg/mL using methanol as a diluent and stored at -40 °C in a freezer (Thermo Fisher Scientific, model ULT2140-5-A40, Waltham, USA). Working standard solutions were freshly prepared every day from the stock solution for each experiment, through appropriate dilution with methanol to achieve concentrations of 5, 7.5, 10, 12.5 and 15 µg/mL (calibration curve).

Sample preparation

For the preparation of the sample solutions, the content of 10 capsules was transferred to a mortar and homogenized with a pestle. The mass corresponding to its medium weight was then dispersed in methanol, using an unfilled 25 mL volumetric flask. The solution was kept in an ultrasonic bath (Branson, model 5510R-DTH, St. Louis, USA) for 15 min at 25 °C. The volume was completed to 25 mL with methanol, and the solution was centrifuged for 4 min at 4000 rpm (Eppendorf, model 5810 R, Hamburg, Germany). Lastly, 0.5 mL was transferred to a 10 mL volumetric flask and completed with methanol, resulting in a theoretical concentration of 10 µg/mL.

Determination of maximum absorption λ_{\max} .

A working standard solution at 10 µg/mL was scanned from 200 to 400 nm with an UV-Vis spectrophotometer and an UV spectrum was obtained. Methanol was used as the blank.

Analytical Validation

Method validation was performed according to Brazilian and international guidelines^{15,16}.

Linearity

Linearity was established from 3 individual analytical curves, prepared by OCL stock solutions, obtaining 5 different concentrations (5, 7.5, 10, 12.5 and 15 µg/ml). The UV absorbance was measured at 287 nm. The limit of detection (LOD) and limit of quantification (LOQ) were calculated as described below:

$$LOD = \frac{3.3 \times \text{standard deviation of regression line}}{\text{slope from regression line}}$$

$$LOQ = \frac{10 \times \text{standard deviation of regression line}}{\text{slope from regression line}}$$

The linearity of the model was evaluated using the least squares regression model. To validate the regression model, the following tests were evaluated: ANOVA, t test, R², Pearson correlation, Durbin-Watson test, homogeneity, and normality of the residues.

Specificity

To survey the specificity of the method, the test was carried out using only excipients. The absorbance of an OCL standard solution (5 µg/mL) was compared with the absorbance of a matrix (excipients) solution composed of a mixture of starch, talc, magnesium stearate, sodium lauryl sulfate and colloidal silicon dioxide. The matrix solution was obtained as a sample preparation. Spectra for the matrix, blank, and sample were also compared. The method was applied to determine if any component of the formulation could generate a response or a read with an absorption band similar to the drug.

Repeatability and Intermediate Precision

The precision was determined according to the repeatability (intra-day) and intermediate precision (inter-day) parameters. Under the same conditions, 6 sample solutions (10 µg/mL) were prepared as described in the "Sample preparation" section and assayed by the UV method at 287 nm (read in triplicate). This procedure was repeated by a different analyst on a non-consecutive day. Sample concentration and relative standard deviations (RSD) were calculated. Intermediate precision data was evaluated by the Shapiro-Wilk test, to evaluate the normality of data distribution ($p > 0.05$), and then the F-test and the Student's t-test.

Accuracy

Accuracy was determined by recovery. Accurately weighed amounts of 4, 5 and 6 mg of OCL reference standard were added to the matrix (excipients) equivalent to a capsule content (130 mg). Then, they were submitted in triplicate to the sample preparation procedure resulting in concentrations of 8, 10 and 12 µg/mL. After UV analysis, the experimental concentrations and relative errors were calculated as described below:

$$ER(\%) = \frac{(measured\ value - nominal\ value)}{nominal\ value} \times 100$$

Robustness

To verify the robustness of the sample preparation method, samples (OCL plus excipients at the nominal dosage) were prepared in triplicate at different ultrasound times (13, 15 and 17 min) and with two different manufacturers of methanol. Robustness was assessed by applying factorial ANOVA to analyze the effect of ultrasound time variation and the Mann-Whitney test was used to evaluate methanol manufacturer variation.

Stability

The stability assay was performed at room temperature (± 25 °C) and under cooling condition (2 °C - 8 °C). The UV-Vis absorbance of the OCL samples (10 µg/mL) and OCL standard solutions (10 µg/mL) was obtained at 0, 24 and 48h for each temperature condition. The Mauchly sphericity (stability) test was applied for stability evaluation.

Statistical analysis

Results were presented as mean \pm standard deviation (SD). Statistical analysis was performed using the SPSS 20 program (USA, New York), and p values <0.05 were considered statistically significant.

RESULTS AND DISCUSSION

A solubility test of OCL with different solvents was previously performed (data not shown). OCL was very slightly soluble in water and acetonitrile, slightly soluble in ethanol and very soluble in methanol. Therefore, methanol was chosen as the solvent for the preparation of the standard and sample.

OCL has conjugated double bonds that are responsible for UV absorption (Figure 1). The UV-Vis spectrum showed that 287 nm was the ideal wavelength; therefore, this wavelength was used for all analyses (Figure 2).

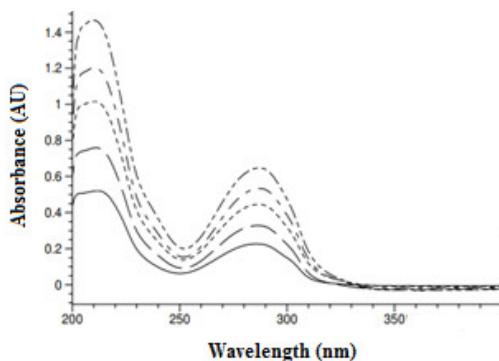


Figure 2. UV spectra of OCL standards at different concentrations

The correlation coefficient (R^2) obtained for the linearity was 0.990, demonstrating that 99% of the concentration variance is predicted by the absorbance of OCL. These values were determined from the mean curve of three analytical curves (Figure 3).

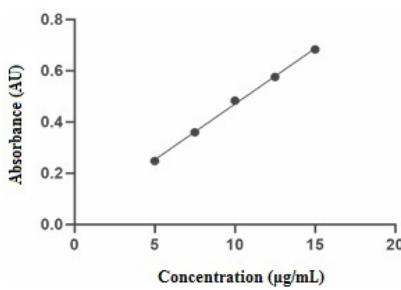


Figure 3. Mean analytical curve of OCL.

Pearson's correlation test was applied between the two variables (concentration and absorbance), both of which were highly correlated ($p = 0.000$ and $r = 0.995$). Then, the quality of the was determined by the ANOVA test and a significant p value ($p = 0.000$) was obtained.

The Student's t-test was also applied to analyze the importance of the regression equation coefficients, in which the linear coefficient was statistically equal to zero ($p = 0.015$). Therefore, the equation was defined by $y = 0.043x + 0.035$.

Then, the Durbin Watson's test was applied to analyze the independence of experimental errors, proving the absence of correlation (value of 1.839, with a maximum acceptable value of 2.5). In addition, the standardized residues had normal distribution (mean = 0 and SD = 1) and presented homogeneity (Figure 4).

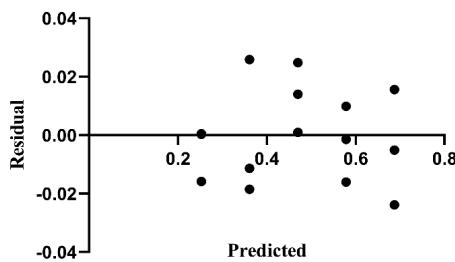


Figure 4. Residual plot from the regression model

Lastly, the limit of detection was found to be 1.18 µg/mL and the limit of quantification was found to be 3.58 µg/mL.

The spectrum for the mixture of excipients (Figure 5) showed no absorption at 287 nm. Thus, the method specificity in the presence of excipients was demonstrated as well as its suitability for the determination of OCL in pharmaceutical formulations.

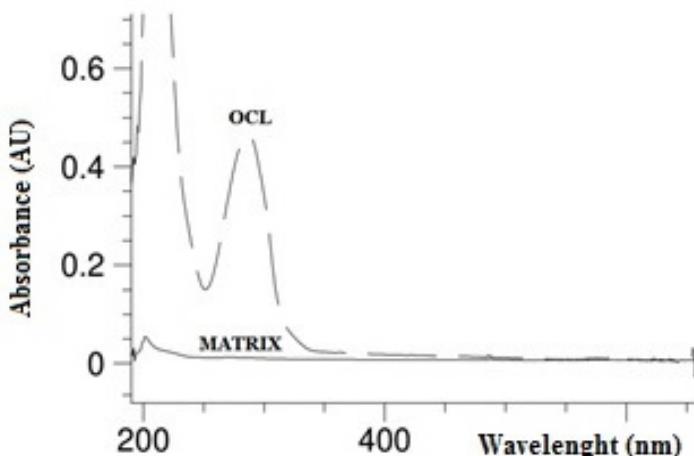


Figure 5. UV-Vis absorbance spectra of the mixture of excipients and OCL standard solution.

The repeatability and intermediate precision results are summarized in Table 1. For both parameters, the values obtained for the RSD were less than 2%, demonstrating adequate repeatability for the analytical method¹⁵. In addition, the Shapiro-Wilk test indicated that the data had a normal distribution ($p = 0.138$), which proves the adequacy of the Student's T test applied. The intermediate precision analysis showed no significant difference between the groups ($p > 0.05$), indicating the precision of the method.

Table 1. Experimental results from repeatability and intermediate precision assays (n=6).

Days	Analyst	Mean		RSD (%)	p-value
		Concentratio n (µg/mL)	SD (µg/mL)		
Day 1	1	11.36	0.22	1.96	0.1405*
Day 2	2	11.26	0.20	1.78	

* no significant difference by Student's t-test. SD, standard deviation; RSD, relative standard deviation.

Accuracy recovery was 98.33%, 100.91% and 98.32% respectively for the 8, 10 and 12 µg/mL concentrations, with a relative error < 2%. This data indicates an appropriate accuracy according to Brazilian guidelines¹⁵.

Robustness was validated by varying the ultrasound time and using different manufacturer of methanol. Neither variations affected the results of samples in concentrations of 10 µg/mL. No significant difference was obtained by using one-way ANOVA ($p = 0.767$) for ultrasound time, or by using Mann-Whitney test ($p = 0.100$) for different methanol manufacturers.

Lastly, the stability of the OCL sample and standards, in cool and at room temperature, was demonstrated for 0, 24 and 48h, with no significant difference between the analyzed times, with p value >0,05 as demonstrated by Table 2.

Table 2. Stability assays of standard and sample solutions of OCL at 10 µg/mL.

	Hours	Mean Concentration (µg/mL)	SD (µg/mL)	p-value
Standard	0	10.10	0.0882	
	24	10.67	0.0825	0.245*
	48	10.55	0.1635	
	0	10.10	0.0882	
	24	9.69	0.0708	0.756*
	48	10.37	0.1661	
Sample	0	10.32	0.2624	
	24	10.43	0.2054	0.521*
	48	10.49	0.2279	
	0	10.67	0.2659	
	24	9.84	0.2272	0.382*
	48	9.84	0.2058	

* Mauchly's Test of Sphericity. SD, standard deviation; RSD, relative standard deviation.

The developed spectrophotometric method was validated according to Brazilian and international guidelines. It is therefore presented as a fast, inexpensive and effective method for quantifying OCL in commercial capsule formulation.

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Authors' contributions

DROM: Conceptualization, Methodology, Investigation, Formal analysis, Validation, Original draft preparation; ROV: Conceptualization, Project administration, Methodology, Investigation, Formal analysis, Original draft preparation; CC: Visualization, Investigation, Formal analysis, Validation, Original draft preparation; FR: Formal analysis, Validation, Original draft preparation, Formal analysis, Validation; KZAD: Writing- Reviewing and Editing, Validation; LSS: Investigation, Review & Editing; AFC: Formal analysis, Review & Editing; RP: Conceptualization, Resources, Original draft preparation, Funding acquisition, Supervision.